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SEARCH REQUEST FORM

Scientific and Technical Information Center

Requester's Full Name: K. C Art Unit: 1913 Phone N Mail Box and Bldg/Room Location:		Examiner #: 770		26/00 KE-MAIL
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Please provide a detailed statement of the search topic, and describe as specifically as possible the subject matter to be searched. Include the elected species or structures, keywords, synonyms, acronyms, and registry numbers, and combine with the concept of utility of the invention. Define any terms that may have a special meaning. Give examples or relevant citations, authors, etc., if known. Please attach a copy of the cover sheet, pertinent claims, and abstract.				
Title of Invention: 5+as, 1,	2 dim 0/-	Polymers	after Expisi	ure to
Inventors (please provide full names):	D. Wang			
Earliest Priority Filing Date:	3/09		A STATE OF THE PARTY OF THE PAR	
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Date Searcher Picked Up: Date Completed: A -27-02	Litigation	Lexis/Nexis		- Maringal Lan.
Searcher Prep & Review Time:	Fulltext	Sequence Systems		
Clerical Prep Time:	Patent Family	WWW/Internet		
Online Time:	Other	Other (specify)		
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FILE 'REGISTRY' ENTERED AT 11:15:06 ON 27 AUG 2002
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=> d his
    FILE 'LREGISTRY' ENTERED AT 10:35:32 ON 27 AUG 2002
               STR
L1
     FILE 'REGISTRY' ENTERED AT 10:36:59 ON 27 AUG 2002
L2
             0 S L1
L3
              SCR 1838 AND 1707 AND 1836 AND 1044
L4
             3 S L1 AND L3
L5
              SCR 2043
             0 S L1 AND L3 AND L5
L6
L7
              STR L1
            2 S L7 AND L3
L8
           469 S L7 AND L3 FUL
              SAV L9 EGW625/A
             0 S L9 AND PMS/CI
L10
    FILE 'LREGISTRY' ENTERED AT 10:46:12 ON 27 AUG 2002
               STR L7
L11
    FILE 'REGISTRY' ENTERED AT 10:49:44 ON 27 AUG 2002
            3 S L11 SSS SAM SUB=L9
L12
L13
            40 S L11 SSS FUL SUB=L9
              SAV L13 EGW625A/A
        39919 S ?BUTYN?/CNS
L14
          32 S L13 NOT L14
L15
L16
            8 S L13 AND L14
    FILE 'CAOLD' ENTERED AT 10:53:25 ON 27 AUG 2002
L17
            0 S L15
    FILE 'ZCAPLUS' ENTERED AT 10:53:34 ON 27 AUG 2002
L18
          169 S L15
               QUE POLYM? OR COPOLYM? OR HOMOPOLYM? OR TERPOLYM? OR RESI
L19
               QUE CROSSLINK? OR CROSS? (2A) LINK?
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L21
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L22
             8 S L18 AND L21
L23
    FILE 'REGISTRY' ENTERED AT 10:56:14 ON 27 AUG 2002
             E C11H14O2
         4575 S E3
L24
L25
           14 S L15 AND L24
L26
           14 S L15 AND (E OR 2E)
L27
           7 S L25 AND L26
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FILE 'ZCAPLUS' ENTERED AT 10:58:24 ON 27 AUG 2002
L28
            49 S L27
L29
           144 S L25
L30
            57 S L26
            10 S (L28 OR L29 OR L30) AND (L19 OR L20 OR L21)
L31
    FILE 'LREGISTRY' ENTERED AT 10:59:29 ON 27 AUG 2002
L32
               STR L11
    FILE 'REGISTRY' ENTERED AT 11:02:14 ON 27 AUG 2002
            0 S L32 SSS SAM SUB=L9
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L34
           10 S L32 SSS FUL SUB=L9
               SAV L34 EGW625B/A
    FILE 'CAOLD' ENTERED AT 11:03:23 ON 27 AUG 2002
             0 S L34
L35
    FILE 'ZCAPLUS' ENTERED AT 11:03:31 ON 27 AUG 2002
L36
           135 S L34
    FILE 'REGISTRY' ENTERED AT 11:03:37 ON 27 AUG 2002
               SEL L34 1-8 RN
L37
             8 S E1-E8
               SEL L34 9 RN
             1 S E9
L38
               SEL L34 10 RN
             1 S E10
L39
   FILE 'ZCAPLUS' ENTERED AT 11:04:48 ON 27 AUG 2002
           96 S L37
L40
            16 S L38
L41
            45 S L39
L42
    FILE 'REGISTRY' ENTERED AT 11:05:09 ON 27 AUG 2002
               SEL L34 1-7 RN
             7 S E11-E17
L43
               SEL L34 8 RN
L44
             1 S E18
    FILE 'ZCAPLUS' ENTERED AT 11:05:39 ON 27 AUG 2002
             4 S L43
L45
            92 S L44
L46
L47
            10 S (L41 OR L42 OR L46) AND (L19 OR L20 OR L21)
            16 S L22 OR L23 OR L31 OR L45 OR L47
L48
L49
            46 S L28 NOT L48
    FILE 'REGISTRY' ENTERED AT 11:15:06 ON 27 AUG 2002
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               SCR 1838 AND 1707 AND 1836 AND 1044
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L7
               STR
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NODE ATTRIBUTES:

CONNECT IS E2 RC AT 7
CONNECT IS E2 RC AT 9
DEFAULT MLEVEL IS ATOM
GGCAT IS SAT AT 7
GGCAT IS UNS AT 9
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 10

STEREO ATTRIBUTES: NONE

L9 469 SEA FILE=REGISTRY SSS FUL L7 AND L3 L11 STR

2 7 CH2 O Ak OH
1 C C 3 8 9 10

NODE ATTRIBUTES:

CONNECT IS E2 RC AT 9
DEFAULT MLEVEL IS ATOM
GGCAT IS UNS AT 9
DEFAULT ECLEVEL IS LIMITED
ECOUNT IS E4 C AT 9

GRAPH ATTRIBUTES:

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NUMBER OF NODES IS 10

STEREO ATTRIBUTES: NONE

L13 40 SEA FILE=REGISTRY SUB=L9 SSS FUL L11

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SEARCH TIME: 00.00.01

40 ANSWERS

=> file zcaplus FILE 'ZCAPLUS' ENTERED AT 11:15:28 ON 27 AUG 2002 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2002 AMERICAN CHEMICAL SOCIETY (ACS)

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=> d 148 1-16 cbib abs hitstr hitind
     ANSWER 1 OF 16
                     ZCAPLUS COPYRIGHT 2002 ACS
              Document No. 136:247398 Etherification method for
2002:220515
     producing benzyl ether derivatives of 2-butene-1,4-diols.
     Krishnamurti, Ramesh; Wang, Qi; Smolka, Thomas F. (Occidental
     Chemical Corporation, USA; Stephens, Dinah, H.). PCT Int. Appl. WO
     2002022537 A2 20020321, 19 pp. DESIGNATED STATES: W: AE, AG, AL,
     AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ,
     DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL,
     IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD,
     MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE, SG, SI,
     SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY,
     KG, KZ, MD, RU, TJ, TM; RW: AT, BE, BF, BJ, CF, CG, CH, CI, CM, CY,
     DE, DK, ES, FI, FR, GA, GB, GR, IE, IT, LU, MC, ML, MR, NE, NL, PT,
     SE, SN, TD, TG, TR. (English). CODEN: PIXXD2. APPLICATION: WO
     2001-GB4036 20010910.
                            PRIORITY: US 2000-PV231790 20000911.
     A method for producing benzyl ethers of 2-butene-1,4-diols, useful
AB
     as polymer stabilizers (no data), comprises
     reacting 2-butene-1,4-diols (e.g., 2-butene-1,4-diol) with benzyl
     chlorides (e.g., benzyl chloride) in the presence of an alkali
     (e.g., sodium hydroxide) or alk. earth metal hydroxide or carbonate,
     optionally in the presence of a phase-transfer catalyst (e.g.,
     benzyltriethylammonium chloride) to produce mono- and/or di-benzyl
     ethers of 2-butene-1,4-diols [e.g., 1,4-bis(benzyloxy)-2-butene] and recovering at least a portion of at least one such benzene deriv.
     80885-30-9P
IT
        (etherification method for producing benzyl ether derivs. of
        2-butene-1,4-diols)
RN
     80885-30-9 ZCAPLUS
     2-Buten-1-ol, 4-(phenylmethoxy)- (9CI)
                                               (CA INDEX NAME)
CN
HO-CH2-CH-CH2-O-CH2-Ph
     I-CM___C07C041-00
IC
```

25-9 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds) CC Section cross-reference(s): 37, 45

70677-94-0P, 1,4-Bis(benzyloxy)-2-butene 80885-30-9P IT (etherification method for producing benzyl ether derivs. of 2-butene-1,4-diols)

L48 ANSWER 2 OF 16 ZCAPLUS COPYRIGHT 2002 ACS
2001:809663 Document No. 136:102461 Stability, Reactivity,
Solution, and Solid-State Structure of Halomethylzinc Alkoxides.
Charette, Andre B.; Molinaro, Carmela; Brochu, Christian
(Departement de Chimie, Universite de Montreal, Montreal, QC, H3C
3J7, Can.). Journal of the American Chemical Society, 123(49),
12160-12167 (English) 2001. CODEN: JACSAT. ISSN: 0002-7863.
Publisher: American Chemical Society.

The development of a Lewis acid-catalyzed cyclopropanation of AB allylic alcs. with bis(iodomethyl)zinc is reported. Iodomethylzinc alkoxides (e.g., chloro- and iodomethyl zinc 4-methoxybenzyloxy compds.) can be formed by treatment of an alc. with bis(iodomethyl)zinc, and their crystal structures solved. These species are not prone to undergo cyclopropanation at low temp., but the addn. of a Lewis acid in catalytic amts. induces the cyclopropanation reaction. Thus, CH2I2 and Et2Zn were stirred and added to cinnamyl alc. in the presence of TiC4, to give the cyclopropanation product, trans-(3-phenylcyclopropyl) methanol, in The Lewis acid-catalyzed pathway significantly overwhelms the uncatalyzed one. The prepn. and stability of halomethyl zinc alkoxides in soln., their aggregation state in soln., and solid-state structures are discussed. Furthermore, the competition reaction between the inter- vs. intramol. cyclopropanation is studied. Finally, the possible activation pathways to explain the Lewis acid activation of halomethylzing alkoxides is discussed.

IT 81028-03-7

(Lewis acid-catalyzed cyclopropanation of allylic alcs. with bis(halomethyl)zinc)

RN 81028-03-7 ZCAPLUS

CN 2-Buten-1-ol, 4-(phenylmethoxy)-, (2Z)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

$$HO$$
 Z O Ph

CC 29-9 (Organometallic and Organometalloidal Compounds) Section cross-reference(s): 22, 75

IT Metal alkoxides

(stability, reactivity, soln., and solid-state structure of halomethylzinc alkoxides)

67-64-1, Acetone, reactions IT 60-12-8, Phenethyl alcohol 75-11-6, 100-44-7, Benzyl chloride, reactions 105-13-5, Diiodomethane 4-Methoxybenzyl alcohol 122-97-4, 3-Phenyl-1-propanol 556-82-1 928-95-0 1576-95-0 4845-04-9, 557-20-0, Diethyl zinc 55131-20-9 1-Cyclohexene-1-methanol 36004-04-3 75553-23-0 81028-03-7 87770-83-0 388568-84-1

(Lewis acid-catalyzed cyclopropanation of allylic alcs. with bis(halomethyl)zinc)

L48 ANSWER 3 OF 16 ZCAPLUS COPYRIGHT 2002 ACS
2001:544578 Document No. 135:272544 One-pot oxidation and wittig olefination of alcohols using o-iodoxybenzoic acid and stable wittig ylide. Maiti, Arup; Yadav, J. S. (Organic Division, Indian Institute of Chemical Technology, Hyderabad, 500007, India). Synthetic Communications, 31(10), 1499-1506 (English) 2001. CODEN: SYNCAV. ISSN: 0039-7911. Publisher: Marcel Dekker, Inc..

AB Benzylic, allylic, and propargylic alcs., as well as diols, can be oxidized with o-iodoxybenzoic acid (IBX) in the presence of stabilized Wittig ylide to generate .alpha.,.beta.-unsatd. ester in one pot. This is useful when the intermediate aldehydes are unstable and difficult to isolate. For example, to a soln. of benzenemethanol in dimethylsulfoxide was added a mixt. of (triphenylphosphoranylidene)acid Et ester and 2-iodoxybenzoic acid. The yield of Et cinnamate thus prepd. was 98%.

IT 81028-03-7

(prepn. of .alpha.,.beta.-unsatd. esters by one-pot oxidn. and Wittig olefination of alcs. using iodoxybenzoic acid and (triphenylphosphoranylidene)acetate (Wittig ylide))

RN 81028-03-7 ZCAPLUS

CN 2-Buten-1-ol, 4-(phenylmethoxy)-, (2Z)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

CC 21-2 (General Organic Chemistry)

IT 100-51-6, Benzenemethanol, reactions 106-24-1 107-19-7,
 2-Propyn-1-ol 107-21-1, 1,2-Ethanediol, reactions 504-63-2,
 1,3-Propanediol 821-11-4 1099-45-2,
 (Triphenylphosphoranylidene)acetic acid ethyl ester 4407-36-7
 4541-14-4 20739-58-6, 2-Octyn-1-ol 64297-64-9, 2-Iodoxybenzoic acid 81028-03-7

(prepn. of .alpha.,.beta.-unsatd. esters by one-pot oxidn. and Wittig olefination of alcs. using iodoxybenzoic acid and (triphenylphosphoranylidene)acetate (Wittig ylide))

ANSWER 4 OF 16 ZCAPLUS COPYRIGHT 2002 ACS
1:508070 Document No. 135:93426 Stabilization of
polymers after exposure to oxidation and article
manufacture. Wang, Qi (USA). U.S. Pat. Appl. Publ. US 20010007884
A1 20010712, 9 pp., Cont.-in-part of U. S. Ser. No. 272,843.
(English). CODEN: USXXCO. APPLICATION: US 2001-761625 20010117.
PRIORITY: US 1999-272843 19990319.
Polyvinyl chloride, polyvinylidene chloride, polycarbonate,

Polyvinyl chloride, polyvinylidene chloride, polycarbonate, polyethylene, polypropylene, polyamide, polyimide, polyether, polyester, or polyvinyl acetate contg. .apprx.0.005-10 phr stabilizer are resistant to discoloring after exposure to

```
oxidn. PVC contq. 1.34 q stabilizer phthalan had initial
     yellowness value 17.5, after .gamma. radiation 31.9, and after aging
     49.5, vs. 18.1, 50.2, and 75.5, resp., for a control without
     stabilizer.
     81028-03-7
IT
        (stabilization of polymers after exposure to
        oxidn.)
RN
     81028-03-7 ZCAPLUS
CN
     2-Buten-1-ol, 4-(phenylmethoxy)-, (2Z)- (9CI) (CA INDEX NAME)
Double bond geometry as shown.
         \mathbf{Z}
                       Ph
HO
IC
     ICM
          C08K005-06
     ICS
          C08K005-46
NCL
     524083000
     37-6 (Plastics Manufacture and Processing)
CC
     discoloration prevention agents Polymer; PVC phthalan
ST
     stabilizer; polyvinylidene chloride yellowing prevention;
     polycarbonate yellowing prevention; polyethylene yellowing
     prevention; polypropylene yellowing prevention; polyamide yellowing
     prevention; polyimide yellowing prevention; polyether yellowing
     prevention; polyester yellowing prevention; polyvinyl acetate
     yellowing prevention
IT
     Antioxidants
     Discoloration prevention agents
     Yellowing prevention (stabilization of polymers after exposure to
        oxidn.)
IT
     Polyamides, uses
     Polycarbonates, uses
     Polyesters, uses
     Polyethers, uses
     Polyimides, uses
     Polyurethanes, uses
        (stabilization of polymers after exposure to
        oxidn.)
                                    332-77-4
                                               496-14-0, Phthalan
IT
     110-65-6, 2-Butyne-1,4-diol
     589-29-7, 1,4-Benzenedimethanol
                                        1606-85-5, 1,4-Bis(2-hydroxy
                                     5417-32-3
                                                  10192-64-0
                                                               21285-46-1
                         2568-24-3
     ethoxy) -2-butyne
                  68972-96-3 81028-03-7
     67461-24-9
        (stabilization of polymers after exposure to
                                              9002-86-2, Poly(vinyl
     9002-85-1, Poly(vinylidene chloride)
IT
                 9002-88-4, Polyethylene
                                             9003-07-0, Polypropylene
     chloride)
     9003-20-7, Poly(vinyl acetate)
        (stabilization of polymers after exposure to
        oxidn.)
```

60.

L48 ANSWER 5 OF 16 ZCAPLUS COPYRIGHT 2002 ACS

2001:155197 Document No. 134:281064 Regioselective synthesis of 13C1-labeled 2-deoxyribonolactones. Hayes, M. P.; Hatala, P. J.; Sherer, B. A.; Tong, X.; Zanatta, N.; Borer, P. N.; Kallmerten, J. (Department of Chemistry, 1-014 Center for Science and Technology, Syracuse University, Syracuse, NY, 13244-4100, USA). Tetrahedron, 57(8), 1515-1524 (English) 2001. CODEN: TETRAB. ISSN: 0040-4020. OTHER SOURCES: CASREACT 134:281064. Publisher: Elsevier Science Ltd..

AB Syntheses of the five regioselectively 13C1-labeled 5-0-benzyl-2-deoxyribonolactones are described. 13C1-Labeled deoxyribonolactones were prepd. by addn. of KCN to epoxides and subsequent lactonization of the resulting nitriles. These mixts. are the pivotal intermediates in the prepn. of 'population labeled' 13C-labeled nucleoside phosphoramidites for solid-phase oligonucleotide synthesis.

oligonucleotide synthesis.
415706-08-0 332865-87-9P 332865-92-6P
332866-00-9P

(regioselective synthesis of 13C1-labeled 2-deoxyribonolactones as synthons for solid phase oligonucleotide synthesis)

RN 415706-08-0 ZCAPLUS

CN 2-Buten-1-ol-2-13C, 4-(phenylmethoxy)-, (2Z)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RN 332865-87-9 ZCAPLUS

CN 2-Buten-1-ol-1-13C, 4-(phenylmethoxy)-, (2E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RN 332865-92-6 ZCAPLUS

CN 2-Buten-1-ol-4-13C, 4-(phenylmethoxy)-, (2E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RN 332866-00-9 ZCAPLUS

CN 2-Buten-1-ol-3-13C, 4-(phenylmethoxy)-, (2E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

CC 33-9 (Carbohydrates)

Section cross-reference(s): 6

IT 415706-03-5 **415706-08-0** 415710-41-7 86633-16-1P 114694-40-5P 332865-86-8P **332865-87-9P** 332865-89-1P 332865-91-5P **332865-92-6P** 332865-94-8P 332865-99-3P **332866-00-9P** 332866-01-0P 332866-04-3P 332875-18-0P 332875-19-1P

(regioselective synthesis of 13C1-labeled 2-deoxyribonolactones as synthons for solid phase oligonucleotide synthesis)

L48 ANSWER 6 OF 16 ZCAPLUS COPYRIGHT 2002 ACS
2000:695906 Document No. 134:17351 Polymer-supported Ti(IV)
and Mn(III) asymmetric alkene epoxidation catalysts. Sherrington,
D. C.; Karjalainen, J. K.; Canali, L.; Deleuze, H.; Hormi, O. E. O.
(Dept. of Pure and Applied Chemistry, University of Strathclyde,
Glasgow, UK). Macromolecular Symposia, 156(IUPAC 8th International
Symposium on Macromolecule-Metal Complexes (MMC-8 Tokyo), 1999),
125-136 (English) 2000. CODEN: MSYMEC. ISSN: 1022-1360. OTHER
SOURCES: CASREACT 134:17351. Publisher: Wiley-VCH Verlag GmbH.

AB Asym. epoxidn. of cis- and trans-allylic and homoallylic alcs. was catalyzed by branched poly(octamethylene-L-(+)-tartrate) and Ti(OiPr)4. Also, Mn(III) macromol. chiral salen complexes operated as catalysts for epoxidn. of alkenes.

IT 81028-03-7

(asym. alkene epoxidn. catalyzed by polymer-supported
Ti(IV) and Mn(III))

RN 81028-03-7 ZCAPLUS

CN 2-Buten-1-ol, 4-(phenylmethoxy)-, (2Z)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

CC 27-2 (Heterocyclic Compounds (One Hetero Atom))

ST alkene asym epoxidn **polymer** supported catalyst

IT Alkenes, reactions

(asym. alkene epoxidn. catalyzed by **polymer**-supported Ti(IV) and Mn(III))

IT Epoxidation

```
Epoxidation catalysts
        (stereoselective; asym. alkene epoxidn. catalyzed by
        polymer-supported Ti(IV) and Mn(III))
     546-68-9, Titanium tetraisopropoxide
                                            135620-04-1
                                                            310405-22-2D,
IT
     polymer_supported 310405-23-3D, polymer
                  310405-24-4D, polymer-supported
     -supported
        (asym. alkene epoxidn. catalyzed by polymer-supported
        Ti(IV) and Mn(III))
IT
     627-27-0, 3-Buten-1-ol
                               763-89-3
                                          771-98-2, 1-Phenyl-1-cyclohexene
                                                               4407-36-7
     928-94-9
                928-95-0 928-96-1
                                     928-97-2
                                                  1576-95-0
     75039-84-8 81028-03-7
        (asym. alkene epoxidn. catalyzed by polymer-supported
        Ti(IV) and Mn(III))
                                                              94992-76-4P
IT
     5775-23-5P
                  17540-04-4P
                                 78513-07-2P
                                               89321-71-1P
     95404-59-4P
                   101976-99-2P
                                   104196-23-8P
                                                   106034-58-6P
                    118759-61-8P
                                    255709-02-5P
     118759-60-7P
        (asym. alkene epoxidn. catalyzed by polymer-supported
        Ti(IV) and Mn(III))
IT
     73784-95-9, uses
        (copolymer; asym. alkene epoxidn. catalyzed by
        polymer-supported Ti(IV) and Mn(III))
     ANSWER 7 OF 16 ZCAPLUS COPYRIGHT 2002 ACS
2000:688304 Document No. 133:267624 Stabilization and
     discoloration prevention of polymers after exposure to
     oxidation and antioxidants therefor. Wang, Qi (Occidental Chemical
     Corporation, USA; Stephens, Dinah). PCT Int. Appl. WO 2000056812 Al
     20000928, 37 pp. DESIGNATED STATES: W: AE, AL, AM, AT, AU, AZ, BA,
     BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB,
     GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC,
     LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT,
     RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM; RW: AT, BE, BF, BJ,
     CF, CG, CH, CI, CM, CY, DE, DK, ES, FI, FR, GA, GB, GR, IE, IT, LU,
     MC, ML, MR, NE, NL, PT, SE, SN, TD, TG. (English). CODEN: PIXXD2.
     APPLICATION: WO 2000-GB247 20000128. PRIORITY: US 1999-272843
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19990319.

GI

$$(R^{2})_{n} \xrightarrow{R^{1}} H \qquad \qquad H \xrightarrow{X} H \qquad H \xrightarrow{X} H \qquad H \xrightarrow{X} H \qquad H \xrightarrow{X} H \qquad H \xrightarrow{X} H \qquad H \xrightarrow{X} H \qquad H \qquad \qquad H \xrightarrow{X} H \qquad H \qquad \qquad H \xrightarrow{X} H \qquad \qquad H \xrightarrow{$$

Poly(vinyl chloride), poly(vinylidene chloride), polycarbonates, AB polyethylene, polypropylene, polyamides, polyimides, polyethers, polyesters, or poly(vinyl acetate) contg. .apprx.0.005-10 phr of a stabilizer I, II, or III, where Z is C, P, Sn, Si, or B; X is -R1C:CR1-, ethynyl, (un) substituted benzene, pyridine, pyrimidine, triazine, furan, pyrrole, thiophene, or naphthalene ring; each Y is independently O, S, or N; each R is independently H, C1-24-alkyl, C6-24-aryl, C7-24-aralkyl; each R1 is independently R, OR, RCO, ROCO, ROCO2, PR2, P(OR)2, PR(OR), NR2, R2NCO, R2NCO2, SR, halogen, or two R1 groups can form a ring; each R2 is independently R, RCO, ROCO, P(OR) 2, Sn(R)q(OR) 3-q, SnRq(OCOR) 3-q, Si(R)q(OR) 3-q, BRq(OR)2-q, or two R2 groups can form a ring; each R3 is independently R, RCO, ROCO, ROCO2, OR, SR, NR2, OPR2, or OP(OR)2; m is 0 when Z is P or B and 1 when Z is Sn, Si, or C; n is 0 when Y is O or S and 1 when Y is N; p is 0-4; and q is 0-3 for the tin stabilizers and 0-2 for the boron stabilizers, with the proviso that in I, when X = -CH:CH-, Y = O and R2 = H, at least one of the R1 is not H; and in II, when X = -R1C:CR1-, Y = Oand Z = C, then at least one of the R3 groups on Z is not H. Thus, a plaque from a compn. comprising PVC 150.00, stearic acid 0.30, Mark 152S 0.23, dioctyl phthalate 97.50, Drapex 6.8 15.00, and phthalan 1.34 q was irradiated with 50 kGy .gamma.-rays, and aged 48 h at 50.degree., giving yellowness 17.5 initially, 31.9 after radiation, and 49.5 after aging, compared with 18.1, 50.2, and 75.5, without phthalan.

81028-03-7, cis-4-Benzyloxy-2-buten-1-ol (stabilization and discoloration prevention of polymers after exposure to oxidn. and antioxidants therefor)

RN 81028-03-7 ZCAPLUS

CN 2-Buten-1-ol, 4-(phenylmethoxy)-, (2Z)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

```
HO
IC
         C08K005-00
     ICM
     ICS
          C08K005-053; C08K005-15; C08K005-06
CC
     37-6 (Plastics Manufacture and Processing)
     antioxidant yellowing prevention polymer; PVC phthalan
ST
     antioxidant gamma radiation; polyolefin antioxidant yellowing
     prevention; vinyl compd Polymer antioxidant yellowing
     prevention; polycarbonate antioxidant yellowing prevention;
     polyurethane antioxidant yellowing prevention; polyamide antioxidant
     yellowing prevention; polyimide antioxidant yellowing prevention;
     polyether antioxidant yellowing prevention; polyester antioxidant
     yellowing prevention
     Yellowing prevention
IT
     Yellowing prevention
        (agents; stabilization and discoloration prevention of
       polymers after exposure to oxidn. and antioxidants
        therefor)
IT
     Antioxidants
        (stabilization and discoloration prevention of
        polymers after exposure to oxidn. and antioxidants
        therefor)
IT
     Polyamides, uses
     Polycarbonates, uses
     Polyesters, uses
     Polyethers, uses
     Polyimides, uses
     Polyurethanes, uses
        (stabilization and discoloration prevention of
        polymers after exposure to oxidn. and antioxidants
        therefor)
IT
     Discoloration prevention agents
     Discoloration prevention agents
        (yellowing; stabilization and discoloration prevention
        of polymers after exposure to oxidn. and antioxidants
        therefor)
                                   110-65-6, 2-Butyne-1,4-diol
     110-64-5, 2-Butene-1,4-diol
IT
                                                496-14-0, Phthalan
     332-77-4, 2,5-Dimethoxy-2,5-dihydrofuran
     589-29-7, 1,4-Benzenedimethanol
                                       1606-85-5, 1,4-Bis(2-
                               2568-24-3, 4,7-Dihydro-2-phenyl-1,3-
     hydroxyethoxy) -2-butyne
                                          67461-24-9
                             21285-46-1
                                                        68972-96-3,
                10192-64-0
     dioxepin
                                    70677-94-0 81028-03-7
     cis-1,4-Bisbenzyloxy-2-butene
     cis-4-Benzyloxy-2-buten-1-ol
        (stabilization and discoloration prevention of
        polymers after exposure to oxidn. and antioxidants
        therefor)
     9002-85-1, Poly(vinylidene chloride) 9002-86-2, PVC
                                                              9002-88-4,
IT
```

Polyethylene 9003-07-0, Polypropylene 9003-20-7, Poly(vinyl acetate)
(stabilization and discoloration prevention of polymers after exposure to oxidn. and antioxidants therefor)

L48 ANSWER 8 OF 16 ZCAPLUS COPYRIGHT 2002 ACS
1999:516268 Document No. 131:257029 Deprotection of tetrahydropyranyl
ethers with montmorillonite K-10 clay in methanol. Taniguchi,
Takahiko; Kadota, Kohei; ElAzab, Adel S.; Ogasawara, Kunio
(Pharmaceutical Institute, Tohoku Univ., Sendai, 980, Japan).
Synlett (8), 1247-1248 (English) 1999. CODEN: SYNLES. ISSN:
0936-5214. OTHER SOURCES: CASREACT 131:257029. Publisher: Georg
Thieme Verlag.

AB A variety of tetrahydropyranyl (THP) ethers are cleaved by montmorillonite K-10 clay in MeOH at room temp. to give the corresponding alcs. THP ethers carrying epoxy, MeOCH2O, Me3CPh2SiO, AcO, and PhCO2 functionalities are stable under the conditions, while ethers carrying ketal, Me3CMe2SiO, and Cl3CC(:NH)O functionalities are unstable.

IT 80885-30-9P

CC

(deprotection of hydropyranyl ethers with montmorillonite K-10 in methanol)

RN 80885-30-9 ZCAPLUS

244298-47-3P

CN 2-Buten-1-ol, 4-(phenylmethoxy)- (9CI) (CA INDEX NAME)

 $HO-CH_2-CH=CH-CH_2-O-CH_2-Ph$

21-2 (General Organic Chemistry)

244781-68-8P

57-88-5P, Cholest-5-en-3-ol (3.beta.)-, preparation 89-78-1P IT 100-02-7P, 4-Nitrophenol, preparation 98-00-0P, 2-Furanmethanol 100-51-6P, Benzyl alcohol, preparation 108-95-2P, Phenol, 111-70-6P, 1-Heptanol 150-76-5P, 4-Methoxyphenol preparation 768-95-6P, Tricyclo[3.3.1.13,7]decan-1-ol 1471-15-4P 4541-14-4P 13482-22-9P 32651-37-9P 35435-68-8P 51326-51-3P 6318-30-5P 80885-30-9P 101992-89-6P 87184-99-4P 95672-91-6P 244298-24-6P 244298-25-7P 244298-46-2P 130372-07-5P

(deprotection of hydropyranyl ethers with montmorillonite K-10 in methanol)

L48 ANSWER 9 OF 16 ZCAPLUS COPYRIGHT 2002 ACS
1998:482469 Document No. 129:189177 Heterogeneous asymmetric
epoxidation of cis-allylic alcohols: use of polymer
-supported Ti(IV)-catalyst. Karjalainen, Jaana K.; Hormi, Osmo E.
O.; Sherrington, David C. (Department of Chemistry, University of
Oulu, Oulu, FIN-90570, Finland). Tetrahedron: Asymmetry, 9(12),
2019-2022 (English) 1998. CODEN: TASYE3. ISSN: 0957-4166. OTHER
SOURCES: CASREACT 129:189177. Publisher: Elsevier Science Ltd..
AB Heterogeneous asym. epoxidn. of cis-allylic alcs. with titanium

isopropoxide and tert-Bu hydroperoxide has been achieved using a branched/crosslinked poly(tartrate ester) ligand. The enantioselectivities and chem. yields obtained are at least comparable to low mol. wt. tartrate ligands.

IT 81028-03-7

(asym. epoxidn. of cis-allylic alcs. catalyzed by **Polymer** -supported Ti(IV) catalyst)

RN 81028-03-7 ZCAPLUS

CN 2-Buten-1-ol, 4-(phenylmethoxy)-, (2Z)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

$$HO$$
 Z O Ph

CC 27-2 (Heterocyclic Compounds (One Hetero Atom))

ST **polymer** supported titanium catalyst asym epoxidn; allylic alc asym epoxidn titanium catalyst

IT Alcohols, reactions

(allyl; asym. epoxidn. of cis-allylic alcs. catalyzed by
polymer-supported Ti(IV) catalyst)

IT Polymer-supported reagents

(asym. epoxidn. of cis-allylic alcs. catalyzed by polymer -supported Ti(IV) catalyst)

IT Epoxidation

Epoxidation catalysts

(stereoselective; asym. epoxidn. of cis-allylic alcs. catalyzed by polymer-supported Ti(IV) catalyst)

IT 546-68-9, Titanium tetraisopropoxide 73784-95-9, uses (asym. epoxidn. of cis-allylic alcs. catalyzed by **polymer** -supported Ti(IV) catalyst)

IT 928-94-9 1576-95-0 **81028-03-7**

(asym. epoxidn. of cis-allylic alcs. catalyzed by **polymer** -supported Ti(IV) catalyst)

IT 78513-07-2P 94992-76-4P 106034-58-6P

(asym. epoxidn. of cis-allylic alcs. catalyzed by **polymer** -supported Ti(IV) catalyst)

L48 ANSWER 10 OF 16 ZCAPLUS COPYRIGHT 2002 ACS

1995:871881 Document No. 124:87505 1-Deoxy-D-xylulose. Kennedy, Isaac A.; Hemscheidt, Thomas; Britten, James F.; Spenser, Ian D. (Dep. Chem., McMaster Univ., Hamilton, ON, L8S 4M1, Can.). Canadian Journal of Chemistry, 73(8), 1329-37 (English) 1995. CODEN: CJCHAG. ISSN: 0008-4042. Publisher: National Research Council of Canada.

1-Deoxy-D-xylulose (1-deoxy-D-threopentulose) is a precursor of thiamin (Vitamin B1) and of pyridoxine (Vitamin B6) in bacteria. The synthesis of a [2,3-13C2] bond-labeled sample of the compd., to be used for investigations of the biosynthesis of the two vitamins, is described. In aq. soln. 1-deoxy-D-xylulose exists mainly as the open chain ketone. In methanol soln. the compd. exists as a mixt. of the open chain ketone and the two corresponding epimeric

furanoses. In acid soln. the compd. yields a dimeric anhydride, di-.beta.-1-deoxy-D-xylulofuranose 2,3':3,2'-dianhydride, whose structure was established by X-ray crystallog.

IT 172510-18-8P

(prepn. of carbon-13 labeled 1-deoxy-D-xylulose)

RN 172510-18-8 ZCAPLUS

CN 2-Buten-1-ol-1,2-13C2, 4-(phenylmethoxy)-, (E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

CC 33-2 (Carbohydrates)

162284-34-6P 162284-35-7P 162287-12-9P IT 60299-43-6P 172510-17-7P 172510-18-8P 172510-20-2P 172510-19-9P 172510-23-5P 172510-24-6P 172510-21-3P 172510-22-4P 172510-25-7P 172510-26-8P 172510-27-9P 172510-29-1P ·172588-95-3P 172588-96-4P 172588-97-5P 172510-30-4P 172588-99-7P 172588-98-6P (prepn. of carbon-13 labeled 1-deoxy-D-xylulose)

L48 ANSWER 11 OF 16 ZCAPLUS COPYRIGHT 2002 ACS

1995:833636 Document No. 123:228823 [2,3-13C2]-4-Hydroxy-L-threonine. Wolf, Eckardt; Spenser, Ian D. (Department of Chemistry, McMaster University, Hamilton, ON, L8S 4M1, Can.). J. Org. Chem., 60(21), 6937-40 (English) 1995. CODEN: JOCEAH. ISSN: 0022-3263. OTHER SOURCES: CASREACT 123:228823.

AB An eight-step synthesis of [2,3-13C2]-4-hydroxy-L-threonine ([2,3-13C2]-(2S,3S)-2-amino-3,4-dihydroxybutanoic acid) is described, starting from [1,2-13C2]acetylene, in an overall yield of 13%. Since a key intermediate of the synthetic sequence, 4-(benzyloxy)-(Z)-but-2-en-1-ol, is available com., the method furnishes a convenient four-step synthesis of nonenriched 4-hydroxy-L-threonine, in an overall yield of 27%.

IT 168292-05-5P

(asym. synthesis of carbon-13 labeled hydroxythreonine from labeled acetylene)

RN 168292-05-5 ZCAPLUS

CN 2-Buten-1-ol-2,3-13C2, 4-(phenylmethoxy)-, (Z)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

34-2 (Amino Acids, Peptides, and Proteins) CC

168292-02-2P, 2-Butyne-1,4-diol-2,3-13C2 168292-04-4P **168292-05-5P** 168292-06-6P IT 168292-03-3P

168292-06-6P 168292-07-7P

168292-08-8P

(asym. synthesis of carbon-13 labeled hydroxythreonine from labeled acetylene)

ANSWER 12 OF 16 ZCAPLUS COPYRIGHT 2002 ACS L48

1993:517385 Document No. 119:117385 .beta.-Phosphorylated five-membered ring nitroxides: synthesis and ESR study of 2-phosphonyl-4-(hydroxymethyl)pyrrolidine aminoxyl radicals. Pierluigi; Finet, Jean Pierre; Le Moigne, Francois; Tordo, Paul (Lab. Struct. React. Especes Paramagn., Univ. Provence, Marseille, 13397, Fr.). J. Org. Chem., 58(16), 4465-8 (English) 1993. ISSN: 0022-3263. OTHER SOURCES: CASREACT 119:117385. JOCEAH.

GI

Intramol. aminomercuration of the alkenyl .alpha.-amino phosphonate AB PhCH2OCH2CH(CMe:CH2)CH2CMe(NH2)P(O)(OEt)2 followed by sodium borohydride redn. leads to the di-Et [4-(benzyloxymethyl)-2,5,5trimethylpyrrolidinyl]phosphonate I (R = PhCH2, X = H). Oxidn. of the phosphonates I (R = H, PhCH2; X = H) with 3-chloroperbenzoic acid led to the stable 2-phosphonylpyrrolidinyl aminoxyl radicals I (same R; X = O.bul.) bearing a 4-(hydroxymethyl) substituent.

81028-03-7 IT

(oxidn. of)

81028-03-7 ZCAPLUS RN

2-Buten-1-ol, 4-(phenylmethoxy)-, (2Z)- (9CI) (CA INDEX NAME) CN

Double bond geometry as shown.

$$HO$$
 Z O Ph

CC 29-7 (Organometallic and Organometalloidal Compounds)
Section cross-reference(s): 22
IT 81028-03-7
(oxidn. of)

L48 ANSWER 13 OF 16 ZCAPLUS COPYRIGHT 2002 ACS
1992:612188 Document No. 117:212188 Total synthesis of
 (+)-10,10-difluorothromboxane A2 and its 9,11 and 15 stereoisomers.
 Witkowski, Stanislaw; Rao, Y. Koteswar; Premchandran, Ramiya H.;
 Halushka, Perry V.; Fried, Josef (Dep. Chem., Univ. Chicago,
 Chicago, IL, 60637, USA). J. Am. Chem. Soc., 114(22), 8464-72
 (English) 1992. CODEN: JACSAT. ISSN: 0002-7863.

An efficient total synthesis of the biol. highly active, AB stable (+)-10,10-difluorothromboxane A2 (I), possessing the abs. configuration of TXA2, was achieved from the chiral synthon II. The key intermediate, the aldehyde III is prepd. in 16 steps with a total yield of 8.8%, which compares with 1.95% in 14 steps by a previously reported chem.-enzymic route. Diastereoselectivity is high in all but one of the steps, the Reformatskii reaction. However, both epimers obtained in this reaction have been converted efficiently to III. The synthesis of the 9.beta.,11.beta. diastereomer IV is also described. As predicted, this geometrically equiv. isomer showed significant binding to the platelet receptor at Kd = 240 nM. It is, however, only a weak agonist causing aggregation of washed human platelets at 0.7% of the activity of I. These data are rationalized in terms of an obligatory H bond between the 9.alpha.,11.alpha.-oxetane O of I and its receptor to achieve

full biol. activity.
108212-54-0

IT

(asym. epoxidn. of)

RN 108212-54-0 ZCAPLUS

2-Buten-1-ol, 4-[(4-bromophenyl)methoxy]-, (Z)- (9CI) (CA INDEX CN NAME)

Double bond geometry as shown.

26-3 (Biomolecules and Their Synthetic Analogs) CC

108212-54-0 IT

(asym. epoxidn. of)

ANSWER 14 OF 16 ZCAPLUS COPYRIGHT 2002 ACS L48 Document No. 109:73876 Irreversible inhibition of papain by epoxysuccinyl peptides. Carbon-13 NMR characterization of the site of alkylation. Yabe, Yuichiro; Guillaume, Dominique; Rich, Daniel H. (Sch. Pharm., Univ. Wisconsin-Madison, Madison, WI, 53706, USA). J. Am. Chem. Soc., 110(12), 4043-4 (English) 1988. CODEN: JACSAT. ISSN: 0002-7863. OTHER SOURCES: CASREACT 109:73876.

GI

E-64 and its analogs rapidly inhibit most cysteine proteinases. AB Although this class of inhibitor is thought to alkylate the sulfhydryl group in the active site cysteine (Cys-25), it has not been known whether the S reacts at C-2 or C-3 of the epoxysuccinyl group in the inhibitor. In order to det. the site of alkylation by cysteine proteinases, the chiral, regiospecifically labeled 13C deriv., EP-475 (I), has been synthesized, and 13C NMR studies of the reaction of I with papain have been done. The 13C NMR spectrum of the complex displayed a new broad signal (line width = 35 Hz) at 76.5 ppm, but no peak near 50 ppm could be detected. A high resoln. difference spectrum, obtained by reaction of activated papain with labeled I and unlabeled EP-475, clearly showed the presence only of the 76.5 resonance. These data establish that the active site thiol group in papain selectively attacks the C-3 carbon of the epoxysuccinyl group in EP-475.

IT 114694-39-2P

(prepn. and asym. epoxidn. of)

RN 114694-39-2 ZCAPLUS

CN 2-Buten-1-ol-2-13C, 4-(phenylmethoxy)-, (E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

CC 34-2 (Amino Acids, Peptides, and Proteins)

IT 114694-39-2P

(prepn. and asym. epoxidn. of)

L48 ANSWER 15 OF 16 ZCAPLUS COPYRIGHT 2002 ACS

1986:108975 Document No. 104:108975 Oxazolidin-2-ones from allylic amines by means of iodine and carbonate anion on polymeric support. A convenient synthesis of (.+-.)-propranolol. Cardillo, Giuliana; Orena, Mario; Sandri, Sergio (Cent. Fis. Macromol., Ist. Chim. "G. Ciamician", Bologna, 40126, Italy). J. Org. Chem., 51(5), 713-17 (English) 1986. CODEN: JOCEAH. ISSN: 0022-3263. OTHER SOURCES: CASREACT 104:108975.

GI

AB A new polymeric reagent was obtained by adsorbing iodine on the resin Amberlyst A 26 in the CO32- form: By treating RCH2CH(NHR1)CH:CH2.HCl (R = Et, PhCH2O, HO; R1 = H, PhCH2) or RCH2CH(NH2)CH:CH2 (R = tetrahydro-2H-pyran-2-yloxy) with this polymeric reagent, (iodomethyl)oxazolidinones I were obtained in very good yield. Iodine displacement, base hydrolysis, and acetylation gave RCH2CH(NHAc)CH(OAc)CH2OAc, while deiodination followed by base hydrolysis and acetylation afforded RCH2CH(NHAc)CHMeOAc, resp. The potential utility of the new polymeric reagent was demonstrated by a short, effective synthesis of (.+-.)-propranolol.

IT 80885-30-9

(addn. reaction with trichloroacetonitrile and thermal rearrangement of)

```
80885-30-9 ZCAPLUS
RN
CN
     2-Buten-1-ol, 4-(phenylmethoxy)- (9CI) (CA INDEX NAME)
HO-CH_2-CH=-CH-CH_2-O-CH_2-Ph
     23-7 (Aliphatic Compounds)
CC
     Section cross-reference(s): 28
ST
     oxazolidinone iodomethyl; allyl amine cyclization carbonate iodine;
     polymer support iodine carbonate; propranolol
     allylisopropylamine conversion
IT
     Iodination
        (cyclization and, of allylic amines with iodine-carbonate anion
        on polymer support)
IT
     Cyclocondensation reaction
        (iodination and, of allylic amines by iodine-carbonate anion on
        polymeric support)
     Polymer-supported reagents
IT
        (iodine-carbonate anion on Amberlyst, for prepn. of
        oxazolidinones from allylic amines)
     Alcohols, preparation
IT
        (amino, prepn. of, by reaction of allylic amines with
        iodine-carbonate anion on polymeric support)
                2305-21-7 80885-30-9
     110-64-5
IT
        (addn. reaction with trichloroacetonitrile and thermal
        rearrangement of)
                   99726-02-0P
                                 99726-03-1P
                                               99726-04-2P
                                                              99726-05-3P
IT
     99726-00-8P
     99726-37-1P
        (prepn. and reaction of, with iodine and carbonate anion on
        polymeric support)
IT
     99726-39-3P
        (prepn. and reaction of, with naphthol anion on polymeric
        support)
IT
     13013-17-7P
        (prepn. of, from allylisopropylamine by reaction with carbonate
        anion-iodine on polymeric support)
IT
     99726-06-4
        (reaction of, with iodine and carbonate anion on
        polymeric support)
     ANSWER 16 OF 16 ZCAPLUS COPYRIGHT 2002 ACS
L48
              Document No. 67:21550 Reaction of xylene dihalides with
1967:421550
             Slezak, Frank B.; Stallings, John P.; Bungs, Janis A.
     diols.
     (Diamond Alkali Co., Painesville, Ohio, USA). Ind. Eng. Chem. Prod.
     Res. Dev., 4, 259-61 From: CZ 1966, (35), Abstr. No. 1113 (English)
            CODEN: IEPRA6.
     1965.
     Treatment of xylylene dihalides with ethylene glycol, trimethylene
AB
     glycol (Ia), and 2-butene-1,4-diol gave a series of ether diols.
     Those from 2,4,5,6-tetrachloro-m-xylylene dichloride (I) and from
     2,3,5,6-tetrachloro-p-xylylene dichloride (II) were used for the
     prepn. of fireproof polyesterstyrene copolymers. Thus,
```

.alpha.,.alpha.' - bis(3 - hydroxypropoxy) - p-xylene, b0.3-0.5 193-200.degree., n25D 1.5178, yield 43%, was prepd. by heating p-xylylene dichloride with Ia and Na2CO3 20 min. at 130-50.degree... Similarly, .alpha.,.alpha.' - bis(4 - hydroxy - 2 - butenyloxy) p-xylene, b0.1 110-13.degree., was prepd. from 2-butene-1,4-diol and 17.5 g. p-xylylene dichloride, yield 9.6 g. -. Also prepd. were 2,3,5,6-tetrachloro-p-xylenes: 31% .alpha.,.alpha.'(bis-2hydroxyethoxy)-, m. 110.5-12.0.degree. (EtCOMe) (diacetate m. 965-97.degree.) from II and ethylene glycol; 4% 1,6-bis(2 hydroxyethoxymethyl - 2,3,5,6-tetrachlorophenyl)-2,5-dioxahexane, m. 170-4.degree. (EtCOMe), diacetate m. 115.5-17.5.degree.. obtained was 99% .alpha.,.alpha.'-bis(4-hydroxy-2-butenyloxy)-pxylylene, m. 99-105.degree., from II and 2-butene-1,4-diol. 2,4,5,6-Tetrachloro-m-xylenes prepd. were: 62% .alpha.,.alpha.'bis(2-hydroxyethoxy)-, m. 100-2.degree. (EtCOMe), from I and ethylene glycol; and 83% .alpha.,.alpha.'-bis(4-hydroxy-2butenyloxy)-, m. 37-40.degree., from 2-butene-1,4-diol and I. 15084-32-9P 15084-33-0P 15084-34-1P

TT 15084-32-9P 15 (prepn. of)

RN 15084-32-9 ZCAPLUS

CN 2-Buten-1-ol, 4,4'-[p-phenylenebis(methyleneoxy)]di- (8CI) (CA INDEX NAME)

$$\label{eq:ch2-o-$$

RN 15084-33-0 ZCAPLUS

CN 2-Buten-1-ol, 4,4'-[(2,3,5,6-tetrachloro-p-phenylene)bis(methyleneoxy)]di- (8CI) (CA INDEX NAME)

$$\begin{array}{c} \text{Cl} \\ \text{CH}_2\text{--O-CH}_2\text{--CH} \\ \text{CH}-\text{CH}_2\text{--O-CH}_2 \\ \end{array}$$

RN 15084-34-1 ZCAPLUS

CN 2-Buten-1-ol, 4,4'-[(2,4,5,6-tetrachloro-m-phenylene)bis(methyleneoxy)]di- (8CI) (CA INDEX NAME)

C1
$$CH_2-O-CH_2-CH=CH-CH_2-OH$$
 $C1$ $C1$ $CH_2-O-CH_2-CH=CH-CH_2-OH$

CC 25 (Noncondensed Aromatic Compounds)

IT 15084-26-1P 15084-27-2P 15084-28-3P 15084-29-4P 15084-30-7P 15084-31-8P 15084-32-9P 15084-33-0P 15084-34-1P

(prepn. of)

=> d 149 1-46 ti

L49 ANSWER 1 OF 46 ZCAPLUS COPYRIGHT 2002 ACS

TI The First Stereoselective Ficini-Claisen Rearrangement Using Chiral Ynamides

L49 ANSWER 2 OF 46 ZCAPLUS COPYRIGHT 2002 ACS

TI First Evidence for the Formation of a Geminal Dizinc Carbenoid: A Highly Stereoselective Synthesis of 1,2,3-Substituted Cyclopropanes

L49 ANSWER 3 OF 46 ZCAPLUS COPYRIGHT 2002 ACS

TI Evidence for the biosynthesis of squalene via the methylerythritol phosphate pathway in a Streptomyces sp. obtained from a marine sediment

L49 ANSWER 4 OF 46 ZCAPLUS COPYRIGHT 2002 ACS

TI First enantioselective non-biological synthesis of asymmetrized tris(hydroxymethyl)methane (THYM*) and bis(hydroxymethyl)acetaldehyd e (BHYMA*)

L49 ANSWER 5 OF 46 ZCAPLUS COPYRIGHT 2002 ACS

TI Novel [2,3]-sigmatropic rearrangement for carbon-nitrogen bond formation

L49 ANSWER 6 OF 46 ZCAPLUS COPYRIGHT 2002 ACS

TI Studies directed toward the syntheses of amphidinolides: formal total synthesis of (-)-amphidinolide P

L49 ANSWER 7 OF 46 ZCAPLUS COPYRIGHT 2002 ACS

TI The asymmetric chelate-Claisen rearrangement as a key step in the syntheses of non-proteinogenic amino acids

L49 ANSWER 8 OF 46 ZCAPLUS COPYRIGHT 2002 ACS

TI Asymmetric epoxide cyclization route to the F-pyran fragment of the

altohyrtins and key aldol studies

- L49 ANSWER 9 OF 46 ZCAPLUS COPYRIGHT 2002 ACS
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- TI A catalytic enantioselective reaction using a C2-symmetric disulfonamide as a chiral ligand: Simmons-Smith cyclopropanation of allylic alcohols by the Et2Zn-CH2I2-disulfonamide system
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- TI Regiochemical control of the ring opening of 1,2-epoxides by means of chelating processes. 10. Synthesis and ring opening reactions of mono- and difunctionalized cis and trans aliphatic oxirane systems
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- TI Asymmetric Cyclopropanation Using New Chiral Auxiliaries Derived from D-Fructose

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- TI A catalytic enantioselective reaction using a C2-symmetric disulfonamide as a chiral ligand: cyclopropanation of allylic alcohols by the diethylzinc-diiodomethane-disulfonamide system
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- TI Enantiodivergent route to the aromatic bisabolane sesquiterpenes via a chiral acetylene alcohol

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- TI Asymmetric construction of optically active 3-hydroxyalkyne functionalities
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- TI Preparation and testing of N-acylamino acid amides as renin inhibitors
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- TI Conversion of 4-oxy-substituted crotonaldehyde into 1-protected (2R)-1,2,4-butanetriol: a new synthetic capacity of bakers' yeast
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- TI Applications of intramolecular amidomercuration. 6. Selectivity in the amination of allylic alcohols via intramolecular amidomercuration
- L49 ANSWER 38 OF 46 ZCAPLUS COPYRIGHT 2002 ACS
- TI Total synthesis of amphoteronolide B and amphotericin B. 1. Strategy and stereocontrolled construction of key building blocks
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- TI Stereocontrolled construction of key building blocks for the total synthesis of amphoteronolide B and amphotericin B
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- TI Stereocontrolled synthesis of 1,3,5...(2n + 1) polyols
- L49 ANSWER 45 OF 46 ZCAPLUS COPYRIGHT 2002 ACS
- TI Synthesis of saccharides and related polyhydroxylated natural products. 1. Simple alditols
- L49 ANSWER 46 OF 46 ZCAPLUS COPYRIGHT 2002 ACS
- TI Factors affecting the endo:exo ratio in Diels-Alder reactions of cyclopentadiene

=> d l41 16 cbib abs hitstr hitrn

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1982:104020 Document No. 96:104020 A mercury mediated route to the
mitosenes. Danishefsky, Samuel; Regan, John (Dep. Chem., Univ.
Pittsburgh, Pittsburgh, PA, 15260, USA). Tetrahedron Lett., 22(40),
3919-22 (English) 1981. CODEN: TELEAY. ISSN: 0040-4039.

GI

AB (E)-PhCH2OCH2CH:CBrCH(OH)CH2CH2OSiMe2CMe3, prepd. in 6 steps from the acetal I, was coupled with 3,2,4-Me(MeO)2C6H2OH under Mitsunobu conditions to give 73% (E)-3,2,4-Me(MeO)2C6H2OCH(CH2CH2OSiMe2CMe3)CB r:CHCH2Ph, which underwent sequential Claisen rearrangement (PhNMe2, 193.degree., 75 min), acetylation (AcCl, Et3N), nitration [Hg(OAc)2, 90% HNO3, Ac2O), and redn. (Zn, AcOH) to give the aniline II. Intramol. cyclocondensation of II [Hg(OAc)2, THF, NaHCO3], followed by silyl cleavage and intramol. cyclization [(EtOCON:)2, Ph3P] gave mitosene III.

II

IT 80885-30-9P

(prepn. and oxidn. of)

RN 80885-30-9 ZCAPLUS

CN 2-Buten-1-ol, 4-(phenylmethoxy)- (9CI) (CA INDEX NAME)

 $HO-CH_2-CH=-CH-CH_2-O-CH_2-Ph$

IT 80885-30-9P

(prepn. and oxidn. of)

=> d 142 45 cbib abs hitstr hitrn

L42 ANSWER 45 OF 45 ZCAPLUS COPYRIGHT 2002 ACS
1979:86836 Document No. 90:86836 Factors affecting the endo:exo ratio in Diels-Alder reactions of cyclopentadiene. Smith, John R. Lindsay; Norman, Richard O. C.; Stillings, Michael R. (Dep. Chem., Univ. York, York, Engl.). Tetrahedron, 34(9), 1381-3 (English) 1978. CODEN: TETRAB. ISSN: 0040-4020.

R R2 R3 III

GI

The ratio of endo-CHO:exo-CHO products in the Diels-Alder addn. of trans-PhCH2OCH2CH:CHCHO (I) to cyclopentadiene (II) can be altered from 1:2 to 9:1 depending on temp. and BF3 catalysis. E.g., for the uncatalyzed reaction, I with II at 0.degree. (240 h) gave a 2:1 mixt. of III (R = CH2OCH2Ph, R1 = R2 = H, R3 = CHO) (IV) to III (R = R3 = H, R1 = CHO, R2 = CH2OCH2Ph) (V) whereas at 200.degree. (6 h), a 1:2 mixt. of IV:V was obtained. For the BF3.Et2O-catalyzed reaction of I and II at 0.degree., a mixt. of .apprx.9:1 IV:V was obtained. The effect of temp. on the uncatalyzed reaction of II with OCHCH:CH(CH2)6CO2H to give III [R = (CH2)6CO2H, R1 = R2 = H, R3 = CHO] and III [R = R3 = H, R1 = CHO, R2 = (CH2)6CO2H] was similar.

(oxidn. of)

RN 69152-88-1 ZCAPLUS

CN 2-Buten-1-ol, 4-(phenylmethoxy)-, (2E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

IT 69152-88-1 (oxidn. of)

=> d 146 92 cbib abs hitstr hitrn

L46 ANSWER 92 OF 92 ZCAPLUS COPYRIGHT 2002 ACS

Document No. 96:143208 Synthesis of saccharides and 1982:143208 related polyhydroxylated natural products. 1. Simple alditols. Katsuki, T.; Lee, A. W. M.; Ma, P.; Martin, V. S.; Masamune, S.; Sharpless, K. B.; Tuddenham, D.; Walker, F. J. (Dep. Chem., Massachusetts Inst. Technol., Cambridge, MA, 02139, USA). J. Org. Chem., 47(7), 1373-8 (English) 1982. CODEN: JOCEAH. 0022-3263.

A new approach to sugar synthesis is demonstrated through syntheses AB of tetritols, pentitols, and hexitols; Ti-catalyzed asym. epoxidn. and new selective opening of 2,3-epoxy alcs. play essential roles. 81028-03-7

IT

60

(asym. epoxidn. of)

81028-03-7 ZCAPLUS RN

2-Buten-1-ol, 4-(phenylmethoxy)-, (2Z)- (9CI) (CA INDEX NAME) CN

Double bond geometry as shown.

$$HO$$
 Z O Ph

81028-03-7 IT

(asym. epoxidn. of)